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Selective synthesis, characterization, and toxicological activity screening of some furan compounds as pesticidal agents

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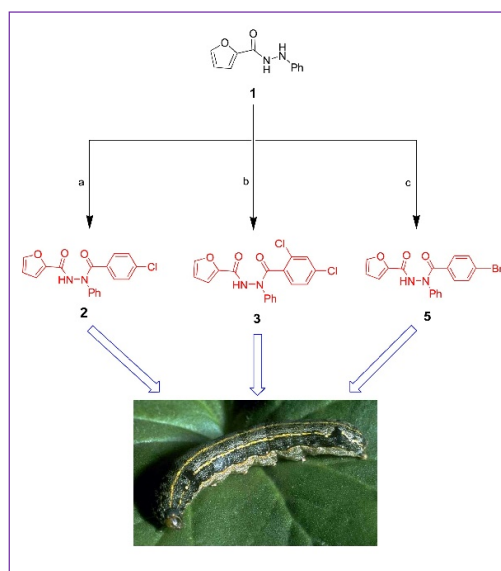
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A series of furan derivatives tebufenozide analogues comprising five compounds (**2**, **3**, **4**, **5**, and **6**) have been designed and synthesized via reaction of *N'*-phenylfuran-2-carbohydrazide (**1**) with different acid chlorides. The structures of the synthesized compounds were identified by elemental analyses and spectroscopic data. The analogues compounds have the active center of tebufenozide (mimic) with change of tertiary butylhydrazine by phenylhydrazine. Studying the effect of this change on the toxicological activity against *Spodoptera littoralis* larvae was performed for three compounds of the synthesized ones. Toxicological activity data showed that compound (**3**) is more active than compounds (**2**) and (**5**).

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Graphical Abstract

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1. Introduction

The importance of synthetic organic compounds, especially furan derivatives, as effective bioactive agents in different fields is well-known. Furan ring, as a critical pharmacophore, is widely present in many active natural products such as toosendanin (a known natural-product insecticidal agents) and limonin.¹⁻⁴ So, different compounds derived from furan were used previously as potential insecticidal agents.^{5,6} Tebufenozide is considered by a lot of scientists to be the safest, most selective, and most useful insect control agent ever to be discovered.⁷ It is a powerful toxicant for controlling both susceptible and field-resistant strains of *S. littoralis*. Tebufenozide compounds have no adverse activity on the predators, *Podisus maculiventris* (Say), *Podisus nigrispinus* (Dallas), and *Orius insidiosus* (Say).^{8,9} Tebufenozide binds to the ecdysteroid receptors by initiating the molting process and force insects to molt prematurely which typically results in stoppage of feeding and ultimately in insect death.¹⁰ Other effects of these compounds on insects include increased egg mortality and reduced rates of reproduction.¹¹ In view of these findings, the current study was planned to synthesize some heterocyclic compounds containing a furan moiety and screening the insecticidal activity against *Spodoptera littoralis* (Boisd.) hoping to find compounds with a good insecticidal activity and being safer toward aquatic life.

2. Results and Discussion

2.1 Chemistry

As following of our project in synthesis and toxicity evaluation of some new tebufenozide (mimic) analogues and the applied fields,¹²⁻²¹ we prepared here some mimic analogues that are shown in **Fig. 1**. The new tebufenozide (mimic) analogues compounds, namely, *N*-phenyl(4-chlorobenzoyl)-furan-2-carbohydrazide **2**, *N*-phenyl(2,4-dichlorobenzoyl)-furan-2-carbohydrazide **3**, *N*-phenyl(4-aminobenzoyl)-furan-2-carbohydrazide **4**, *N*-phenyl(4-Bromobenzoyl)-furan-2-carbohydrazide **5**, and *N*-phenyl(thenoyl)-furan-2-carbohydrazide **6** were synthesized as follow:

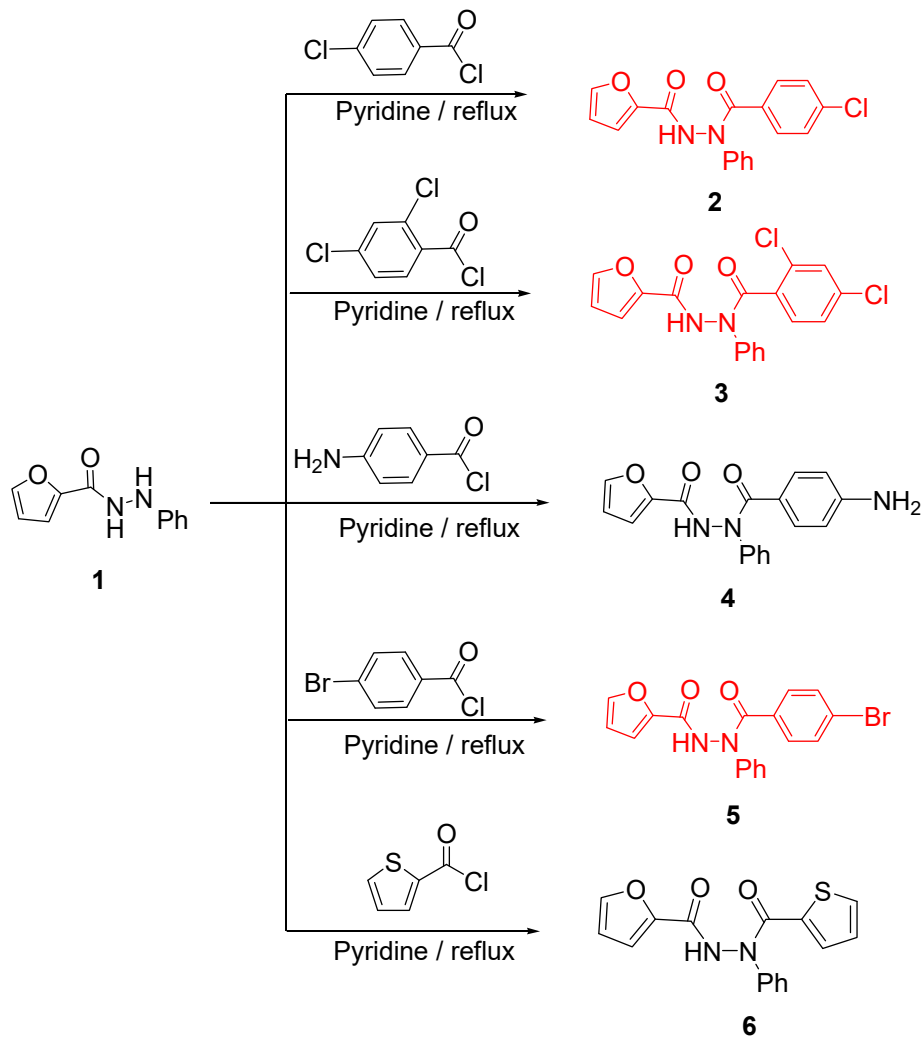


Fig. 1. Synthesis of compounds **2**, **3**, **4**, **5** and **6**

N-Phenyl furan-2-carbohydrazide (**1**) reacts with different acid chlorides namely; 4-chlorobenzoyl chloride, 2,4-dichlorobenzoyl chloride, 4-amino benzoyl chloride, 4-bromobenzoyl chloride and thenoyl chloride, by refluxing in pyridine to give: *N*-phenyl(4-chlorobenzoyl)-furan-2-carbohydrazide **2**, *N*-phenyl(2,4-dichlorobenzoyl)-furan-2-carbohydrazide **3**, *N*-phenyl(4-aminobenzoyl)-furan-2-carbohydrazide **4**, *N*-phenyl(4-Bromobenzoyl)-furan-2-carbohydrazide **5** and *N*-phenyl(thenoyl)-furan-2-carbohydrazide **6**.

The structure of all newly synthesized compounds were elucidated and confirmed on the basis of their elemental analyses and spectroscopic data. The results of elemental analyses were found to be in good agreement with the calculated values. The spectral data of all prepared compounds were in accordance with their proposed structures.

IR spectrum for compound **2** showed absorption bands at 3140 and 1673 cm^{-1} characteristics for (NH) and (C=O) groups respectively. ^1H NMR (DMSO- d_6) spectrum for compound **2** showed that NH group at 12.5 ppm, CH aromatic proton signals at 7.97-7.59 ppm. ^{13}C NMR (DMSO- d_6) spectrum of compound **2** showed different signals at 168.4 (C=O), 163.6 (C=O), 151.0 (C-Cl), 149.3 (C-N), 142.5 (C-CO), 140.3 (C-CO), other aromatic C-H carbons at 131.8, 120.9, 120.1, 119.5, 117.2, 116.9, 115.3, 112.4.

2.2 Toxicity test

Toxicity test for the 2nd instar larvae of the cotton leaf worm *S. littoralis* (Boisd.), revealed compound **3** is the most effective compound, giving LC_{50} value (269 ppm.), while compounds **2** and **5** showed the LC_{50} values (397 and 499 ppm), respectively (Table 1).

Table 1. Toxicity of compounds **2**, **3** and **5** against 2nd instar of *S. littoralis*

Compound	LC_{25} (ppm)	LC_{50} (ppm)	Slop
2	170	397	1.830 +/- 0.711
3	93	269	1.466 +/- 0.717
5	207	499	1.768 +/- 0.696

Toxicity test for the 4th instar larvae of the cotton leaf worm *S. littoralis* (Boisd.), revealed compound **3** is the most effective compound, giving LC_{50} value (496 ppm.), while compounds **2** and **5** showed the LC_{50} values (787 and 660 ppm), respectively (Table 2).

Table 2. Toxicity of compounds **2**, **3** and **5** against 4th instar of *S. littoralis*

Compound	LC_{25} (ppm)	LC_{50} (ppm)	Slop
2	327	787	1.772 +/- 0.699
3	115	496	1.064 +/- 0.671
5	330	660	2.243 +/- 0.717

3. Conclusion

This work involves the synthesis of five heterocyclic compounds (**2**, **3**, **4**, **5** and **6**) and studying the biological activity as potential insecticides for compounds (**2**, **3**, and **5**) against 2nd and 4th instar of *S. littoralis*. The results showed that the compound **3** is more active than the compounds **2** and **5**. This result is encourageable for further work searching on new insecticidal agents.

4. Experimental

4.1 Materials and methods

Melting points are uncorrected and were determined by Kofeler melting point apparatus. IR (cm^{-1}) spectra were listed (KBr disc) on a Shimadzu DR-8001 spectrophotometer. ^1H NMR and ^{13}C NMR (DMSO- d_6) spectra were listed at 400 and 100 MHz on a Varian Gemini NMR spectrometer. The chemical shift is expressed in δ value (ppm) using TMS as an internal reference. Elemental analyses were carried out on a Perkin-Elmer 240°C Micro analyzer.

The present work was conducted to prepare new derivatives of Tubofenozide (mimic)

4.2 Synthesis of compounds 2, 3, 4, 5 and 6.

4.2.1 General procedures:

Freshly prepared acid chloride (1mol) was added drop wise while stirring to an equimolecular amount of *N'*-Phenylfurancarbohydrazide (1mol) in 20 ml pyridine and refluxed for 8-16 hrs. A solution of hydrazide derivative was poured on ice water acidified by HCl. The resulting precipitate was collected by filtration, washed thoroughly with H₂O and purified by crystallization from methanol.

4.2.2 *N'*-phenyl (4-chlorobenzoyl)-furan-2-carbohydrazide (2):

Brown solid (73% yield); mp. 158-160°C; IR (ν , cm⁻¹): 3140.2 (NH), 3045.8 (CH_{arom}), 1673.0 (C=O). ¹HNMR (DMSO-*d*₆), (δ ppm): 12.5 (s, 1H, NH_{exch}), 7.97-7.59 (m, 12H, H_{arom}). ¹³CNMR (DMSO-*d*₆), (δ ppm): 168.4 (C=O), 163.6 (C=O), 151.0 (C-Cl), 149.3 (C-N), 142.5 (C-CO), 140.3 (C-CO), other aromatic C-H carbons at 131.8, 120.9, 120.1, 119.5, 117.2, 116.9, 115.3, 112.4. *Anal.* For C₁₈H₁₃ClN₂O₃ (340.76): Calcd./found C: 63.44/63.21, 3.85/3.88, 8.22/8.32.

4.2.3 *N'*-phenyl (2, 4 dichlorobenzoyl)-furan-2-carbohydrazide (3):

Brown solid (83% yield); mp. 147-150°C; IR (ν , cm⁻¹): 3209.2 (NH), 3045.8 (CH_{arom}), 1669.7 (C=O). ¹HNMR (DMSO-*d*₆), (δ ppm): 12.4 (s, 1H, NH_{exch}), 7.96-7.68 (m, 11H, H_{arom}). ¹³CNMR (DMSO-*d*₆), (δ ppm): 168.4 (C=O), 163.6 (C=O), 153.0 (C-Cl), 152.0 (C-Cl), 149.3 (C-N), 142.5 (C-CO), 140.3 (C-CO), other aromatic C-H carbons at 128.4, 122.7, 120.3, 119.8, 118.9, 118.6, 117.9, 114.2, 112.9. *Anal.* For C₁₈H₁₂Cl₂N₂O₃ (375.2): Calcd./found C: 57.62/57.53, 3.22/3.33, 7.47/7.69.

4.2.4 *N'*-phenyl (4-aminobenzoyl)-furan-2-carbohydrazide (4):

Brown solid (80% yield); mp. 130-134°C; IR (ν , cm⁻¹): 3166.2 (NH), 3092.6 (CH_{arom}), 1658.6 (C=O). ¹HNMR (DMSO-*d*₆), (δ ppm): 12.6 (s, 1H, NH_{exch}), 7.91-7.43 (m, 12H, H_{arom}), 5.6 (s, 2H, NH_{exch}), ¹³CNMR (DMSO-*d*₆), (δ ppm): 166.4 (C=O), 163.9 (C=O), 148.1 (C-N), 144.6 (C-NH₂), 142.4 (C-CO), 141.6 (C-CO), other aromatic C-H carbons at 128.7, 123.9, 121.1, 119.2, 118.2, 116.9, 112.4, 112.2. *Anal.* For C₁₈H₁₅N₃O₃ (321.33): Calcd./found C: 67.28/67.30, 4.71/4.87, 13.08/13.10.

4.2.5 *N'*-phenyl (4-Bromobenzoyl)-furan-2-carbohydrazide (5):

Brown solid (77% yield); mp. 108-112°C; IR (ν , cm⁻¹): 3194.2 (NH), 3086.64 (CH_{arom}), 1655.30 (C=O). ¹HNMR (DMSO-*d*₆), (δ ppm): 11.95 (s, 1H, NH_{exch}), 7.91-7.439 (m, 12H, H_{arom}). ¹³CNMR (DMSO-*d*₆), (δ ppm): 168.2 (C=O), 166.9 (C=O), 154.9 (C-Br), 149.3 (C-N), 140.4 (C-CO), 139.6 (C-CO), other aromatic C-H carbons at 130.8, 122.3, 120.5, 118.5, 117.9, 116.9, 112.8, 111.7. *Anal.* For C₁₈H₁₃BrN₂O₃ (385.21): Calcd./found C: 56.12/56.25, 3.40/3.36, 7.27/7.33.

4.2.6 *N'*-phenyl(thenoyl)-furan-2-carbohydrazide (6):

Brown solid (73% yield); mp. 133-134°C; IR (ν , cm⁻¹): 3140.2 (NH), 3074.40 (CH_{arom}), 1673.15 (C=O). ¹HNMR (DMSO-*d*₆), (δ ppm): 12.51 (s, 1H, NH_{exch}), 7.97-7.59 (m, 12H, H_{arom}). ¹³CNMR (DMSO-*d*₆), (δ ppm): 167.3 (C=O), 166.2 (C=O), 149.3 (C-N), 143.9 (C-CO), 142.3 (C-CO), other aromatic C-H carbons at 133.5, 128.6, 124.6, 122.3, 122.1, 119.3, 117.8, 107.9, 105.3. *Anal.* For C₁₆H₁₂N₂O₃S (312.34): Calcd./found C: 61.53/61.55, 3.87/3.72, 8.97/8.89.

4.3 Biological tests

4.3.1 Toxicological studies.

The present work was conducted to study the susceptibility in laboratory of 2nd and 4th instars larvae of the cotton leaf worm *S. littoralis* (Boisd.) to the Mimic derivatives.

4.3.2 Laboratory bioassay (*S. littoralis*).

From pretest furan derivatives **2**, **3** and **5** were the highest active compounds, so they chosen for bioassay test. A series of concentrations (acetone) for each compound were prepared as the active ingredients based on ppm by diluting with water. Castor-bean leaves were dipped (Leaf-dipping technique corresponding to that described by Tabashnik)²² for 30 seconds in each concentration then left to dry for one hour. The 2nd and 4th instars larvae tested were confined with treated leaves in glass jars covered with muslin for 24 hrs. Treated leaves were then removed and fresh untreated leaves provided. Three replicates (each of 20 larvae) were tested for each concentration. The average of mortality percentage was corrected using Abbott's formula.²³ The corrected mortality percentage of each compound was statistically computed according to Finney, (1971).²⁴ From which the corresponding concentration probit lines (ld-p lines) were estimated in addition to determine 50

and 90% mortalities, slope values of tested compounds were also estimated. This work ensures the importance of agricultural applications in life.²⁵⁻²⁶

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